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RESEARCH ARTICLE

PHYTOCHEMICAL STUDIES AND *IN VIVO* ANTIOXIDANT ACTIVITY OF TWO *LAVANDULA* SPECIES (LAMIACEAE) AGAINST STREPTOZOTOCIN INDUCED OXIDATIVE STRESS IN ALBINO RATS

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ABSTRACT

Phytochemical studies of two *Lavandula* species namely *L. dentata* L. and *L. angustifolia* Mill. (Lamiaceae) cultivated in Egypt was carried out to subsist positive diagnostic indices for the research of their monographs as well as GC/MS analysis of their volatile oils and quantitative estimation of total phenol and flavonoid contents. The gained results revealed both qualitative and quantitative variation in the chemical composition of the

investigated species where twenty-six and thirty-one compounds were identified in the volatile oils of *L. dentata* L. and *L. angustifolia* Mill. representing 96.43 % and 98.56 % respectively, the major chemical constituents in the volatile oil of *L. dentata* were menthe-1,5-dien-8-ol $\langle Para \rangle$ (26.80%), caryophyllene oxide (16.40%) and guiaol $\langle Alpha \rangle$ (15.36%), while in *L. angustifolia* were linalyl acetate (18.99%), citronellol $\langle Alpha \rangle$ (17.36%) and menthe-1,5-dien-8-ol $\langle Para \rangle$ (26.80%), citronellol $\langle Alpha \rangle$ (17.36%) and menthe-1,5-dien-8-ol $\langle Para \rangle$ (16.21%), moreover, the calculated values of total phenol and flavonoid contents were 188.50±2.07 mg GAE g⁻¹, 90.40±1.57 mg QE g⁻¹, 194.95±2.55 mg GAE g⁻¹, 98.83±2.68 mg QE g⁻¹ and 167.10±2.30 mg GAE g⁻¹, 116.15±1.91QE g⁻¹,152.50±2.21mg GAE g⁻¹, 119.95±1.87 mg QE g⁻¹ for their aqueous and methanol extracts respectively.

Investigation of their antioxidants impact against Streptozotocin-induced oxidative stress in liver and kidney tissues of albino rats revealed that the methanol extract of *L. angustifolia* exhibited the highest antioxidant potential followed by, that of *L. dentata* and volatile oil *L. angustifolia* in dose dependent manner respectively which afford useful results for developing new natural antioxidant agents.

Key words: Lavandula, Volatile Oils, Phenols, Flavonoids, Antioxidant, Streptozotocin, Glibenclamide and Oxidative Stress.

INTRODUCTION:

Genus Lavandula (Lamiaceae), consist of about 39 species, dozens of subspecies, hundreds of hybrids and cultivars those are widely distributed in the archipelagoes of the Atlantic Ocean and Mediterranean region, it is divided into four main categories; *L. angustifolia*, *L. dentata*, *L. latifolia* and *L. intermedia* (1 and 2), in Egypt, there are two Lavandula species; *L. dentata* L. and *L. angustifolia* Mill. (3); *L. dentata* commonly known as French lavender is a large plant with greenish-grey foliage late blooming and with characteristic very strong odour (4) while *L. angustifolia* commonly known as English lavender is a frost hardy species that has many cultivars, habitats and blossom colours (5).

The advantageous value of *Lavandula* species can be referred to the virtue of its versatile therapeutic potentials those can be attributed to the high content of their uniquely constituted volatile oil as well as phenolic

content (1, 2 and 6); several researches reported isolation of over 150 compounds from the volatile oil of *L. dentata* (7) while that of *L. angustifolia* contained at least 38 different compounds, the chemical compositions of these volatile oils are complex and variable as reported in several investigations for cultivated species in Saudi Arabia, Spain, Morocco, Algeria and Canada (4, 8-20), table, 1.

The most prominent chemical constituents in the volatile oil *L. dentata* (8) are1,8-cineole, α -pinene, β -pinene and pcymene which exhibit antifungal and antibacterial activities (21 and 8), α -terpineol, terpenen-4-ol and camphene have anti-lice activity (22), whereas those in the volatile oil of *L. angustifolia* (7 and 23) are linally acetate and linalool which have sedative (24 and 25) and local anesthetic effects (26); linalool also exhibited antibacterial (5 and 27-29), fungistatic (21, 27 and 30-32) and insecticidal (22) effects.

Spp.	Major constituents	Uses	Country	Reference (s)
L.	Camphor, <i>trans</i> -pinocarveol, β -eudesmol and α -guiaol.	Antimicrobial and antioxidant	Yemen	9
ntata	β -pinine, pinocarveol, myrtenal, α -pinene and 1, 8 cineol.	Cytotoxic and antioxidant	Morocco	10
Lavandula dentata L.	1, 8 cineol, sabinene, bicycle [3.1.0] hex- an-3-ol,4-methylene-1-(1-methylethyl), myrtenal and α -pinene.	Antimicrobial	Morocco	4
Lavan	1, 8-cineole, cis-verbenol, p-cymen-8-ol, fenchone and myrtenal.		Algeria	11and 12
	β -pinine, pinocarveol and myrtenal		Spain	13
I .	Caryophyllene, β -phellandrene and eucalyptol.	Antimicrobial	Romania	14
ia Mil	Linalyl acetate, linalool, lavandulol, 1, 8- cineole, lavandulyl acetate and camphor.		Canada	15
stifolı	Linalool, camphor, linalyl acetate, (Z)-B- ocimine, 1, 8-cineole and (E)-β-ocimene.	Antioxidant	Iraq	16
Lavandula angustifolia Mill.	Linalool, linalyl acetate, thujene and α -pinene.		Iran	17
lulc	1, 8-cineole, camphor and linalool.		Italy	18
avana	Linalyl acetate, linalool, lavandulol, 1, 8- cineole, lavandulyl acetate and camphor.	Cytotoxic	India	19
Γ	Linalyl acetate, linalool, lavandulol, 1, 8- cineol, lavandulyl acetate and camphor.	Antimicrobial	USA	20

Table 1: The reported major constituents and biological activities of the volatile oils isolated from Lavandula spp.
under investigation:

Free radicals (reactive oxygen and nitrogen species, ROS/RNS) are produced in normal and pathological cell metabolism and proved to be aggravated in cases of oxidative stress; they are controlled by endogenous enzymes as superoxide dismutase, glutathione peroxidase, catalase or chemical compounds as α -tocopherol, ascorbic acid, phenol compounds and glutathione (**33**).

Phenolic compounds are a class of secondary metabolites existing in most plants protecting them against ultraviolet radiation, pathogens and herbivores; concerning phenolic compounds in the investigated species, previous studies reported the existence of flavonoids as genkwanin, luteolin, apigenin, apigenin-7-O-glucoside , apigenin-7-O-glucuronide in both species whereas, 6-hydroxyluteolin-7-O-glycoside, scutellarein-7-O-glycoside, vitexin and luteolin 7-O-gulcour-inonide, luteolin 7,4`-O-di-O-glucuronide, luteolin-7-O-glycoside and chrysoeriol-7-O-glycoside existed in *L. dentata* and *L. angustifolia* respectively (**34** and **35**).

Phenolic compounds exhibit a wide range of pharmacological activities depending on their ability to inhibit certain enzymes meanwhile induction of other detoxifying and antioxidant enzymes, so they revealed to be useful treatment and/or prophylaxis of certain diseases as different inflammations, Alzheimer's disease (**36** and **38**), cancers (**38** – **40**), bacterial, protozoal and viral infections (**39**, **40**, **42** and **43**), heart diseases, stroke, arteriosclerosis, diabetes (**14**, **15** and **44**).

In the last years, there is an increasing interest in antioxidants and the preference of natural rather than synthetic sources (**39-41**) hence, the interest in the antioxidant potential of plant extracts, essential oils and/or isolated compounds, so the present study was conducted to lay a phytochemical foundation for assessment of the volatile oils, phenol and flavonoid contents of two Egyptian *Lavandula* species monitored with investigation of their *in vivo* antioxidant potentials.

MATERIAL AND METHODS:

Plant material:

Shrubs of *Lavandula dentata* L. were purchased from an Ornamental Plants Cultivation Station, Kerdasa Road, Giza while those of *Lavandula angustifolia* Mill. was purchased from The Medicinal Plants Cultivation Station, Faculty of Pharmacy, Cairo University, on April 2013, their identities were established by Professor Dr. Mohamed El-Sayed Tantawy, Professor of Botany, Faculty of Science, Ain Shams University. Voucher specimens are kept in a herbarium in Pharmacognosy Department, Faculty of Pharmacy Al-Azhar University, Cairo, Egypt; plant material were air-dried, powdered and kept in tightly closed amber coloured glass containers protected from light at low temperature.

Chemicals:

Folin-Ciocalteu's reagent, 1,1-Diphenyl-2-picryl-hydrazil (DPPH) and Streptozotocin (Sigma Chemical Co., Saint Louis, MO, USA); gallic acid, quercetin, aluminum chloride and Silica gel 60 F_{254} (E. Merck, Darmstadt, Germany), Glibenclamide (Daonil®, Hochest Company for Pharmaceuticals, Cairo, A.R.E), solvents of analytical grade and pre-made kits suitable for automatic biochemical analyzers for evaluation of blood glucose level (BGL), superoxide dismutase (SOD), reduced glutathione (GSH), catalase (CAT) in RBC_S and evaluation of serum ascorbic acid (Randox Laboratories, Crumlin, England).

Determination of physicochemical parameters (45); ash values; total ash, water-soluble ash and acid-insoluble ash values, moisture content and solvent extractives' values were determined for each of the powdered plants.

Investigation of volatile oils (45):

- **Preparation of the volatile oils;** fresh entire aerial parts were comminuted and separately hydro-distilled in Clevenger's system, the resulted volatile oils were dried over anhydrous sodium sulfate and kept in a refrigerator until analysis.

- Determination of percent yields, specific gravities and *refractive indices;* the percent yields, specific gravities and refractive indices were determined for both isolated oils.

- *GC/Ms analysis of volatile oils;* Thermo Scientific capillary gas chromatography (model Trace GC ULTRA) directly coupled to ISQ Single Quadruple MS, TG-5MS non-polar 5% Phenyl Methyl poly-siloxane capillary column ($30m \ge 0.25mm$ ID $\ge 0.25um$) system was used where programmed analysis was first carried out; the oven temperature programmed from 40°C ($3 \min$) to 280°C at5°C/min, then isothermal at 280 °C for 5 min; carrier gas Helium, flow rate 1mL/min; the volume of injected sample was 1µl of sample in diethyl ether; splitless injection technique; ionization energy 70eV, in electronic ionization (EI) mode.

Preparation of extracts; for each plant, 50 g each finelypowdered sample were extracted separately in a soxhlet with 250 ml of 80% (v/v) aqueous methanol and distilled water for one hour to yield the methanol and aqueous extracts respectively, the extracts were filtered under vacuum through Whatmann No.1 filter paper, the residue was re-extracted following the same procedure two more times, extracts collected were vacuum dried at 40° C.

Determination of total phenol content (46); the total phenol content in aqueous and methanol extracts of each plant was determined spectrophotometrically using Folin-Ciocalteu's reagent where standard curve was done using different concentrations of gallic acid in methanol. The concentrated extracts of tested plants were dissolved each in least methanol volume then completed to 10ml, 100µl of these extracts were separately diluted with 8ml distilled water, to each sample 0.5ml of 50% Folin-Ciocalteu's reagent was added and left 8 min, and then 1.5 ml of 5% sodium carbonate were added, mixed and allowed to stand for 60 min. protected from light. The absorbance was measured using Genesys Spectrophotometer (Milton Roy, INC., Rochester, NY) at 725 nm where methanol was used as blank and concentration of the total phenolic content was calculated as mg gallic acid equivalents per gram dry weight (mg GAE g⁻¹).

Determination of total flavonoid content (47); the total flavonoid content in in aqueous and methanol extracts was done colourimetrically using aluminum chloride solution where standard curve was done using different concentrations of quercetin in methanol, 100µl were added to a 96 Micro-well plate, 100µl of 2% aluminum chloride solution in methanol were added, after 10 min, their absorbance was measured using Genesys Spectrophotometer (Milton Roy, INC., Rochester, NY) at 415 nm where methanol was used as blank and the concentration of total flavonoids was calculated as mg quercetin equivalent per gram dry weight (mg QE g⁻¹).

Preliminary screening of antioxidant potential (48); the tested extracts and volatile oils were subjected to qualitative TLC investigation according to the stable DPPH radical technique where 20 μ l aliquot of each extract was spotted on silica gel plates and developed using *n*-butanol: acetic acid: water (4:1:5) as a mobile phase, after development, the dried TLC plates were sprayed with 0.2% DPPH solution in methanol and examined after 30 min. where active antioxidants appeared as yellow spots against purple background.

Biological Studies:

Acute toxicity (49); adult albino mice of either sex were fasted overnight, divided into six groups-six animals each (1:1 for males: females) - were fed with increasing doses of isolated volatile oils and methanol extracts 50, 100, 200, 500, 1000 and 2000 mg/ kg b.wt. where animals were observed for any toxic symptoms and mortalities during the first 24-72 hours post treatment.

Evaluation of antioxidant potential; adult albino rats of either sex weighing 150-200 g purchased from The Animal House Laboratory, National Research Center, Cairo, Egypt were used for this study, they were housed in an environmentally control room, maintained at uniform light and temperature conditions of and provided with

food and water *ad libitum*.

Induction of diabetes; rats were rendered diabetic by a single intraperitonial dose of freshly prepared Streptozotocin 45 mg kg⁻¹ b.wt. dissolved in saline where diabetes was identified in rats by moderate polydipsia and marked polyuria, 48 h., later fasting blood glucose levels were estimated, rats with blood glucose levels ranging between 200-350 mg dl⁻¹ were considered diabetic and included in the experiment.

Experimental design; one hundred diabetic rats were divided into ten groups - ten animals each (1:1 for males: females) - as follows; **group-1;** received standard saline solution (10 ml kg⁻¹b.wt.),**group-2;** received the reference standard Glibenclamide (0.025 g kg⁻¹ b.wt.), **group-3;** received *L. dentata* volatile oil (50 mg kg⁻¹ b.wt.),**group-4;** received *L. dentata* volatile oil (100 mg kg⁻¹b.wt.), **group-5;** received *L. angustifolia* volatile oil (50 mg kg⁻¹b.wt.), **group-6;** received *L. angustifolia* volatile oil (100 mg kg⁻¹b.wt.); **group-7;** received *L. dentata* methanol extract (50 mg kg⁻¹b.wt.), **group-9;** received *L. dentata* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), **group-9;** received *L. angustifolia* methanol extract (100 mg kg⁻¹b.wt.), as well as **group-11** which was formed of normal

untreated animals received standard saline solution (10 ml kg^{-1} b.wt.).

All the investigated volatile oils and extracts were given as single daily oral doses; they were sacrificed on 30^{th} day by cervical dislocation and the gained liver and kidney tissues' homogenates were used for estimation of enzymatic antioxidants such as superoxide dismutase (50) and catalase (51) and non-enzymatic antioxidants such as vitamin C (52) and glutathione (53).

RESULTS AND DISCUSSION:

The physiochemical parameters determined for the powdered aerial parts of *L. dentata* L. and *L. angustifolia* Mill. are compiled in table (2).

 Table 2: Some physiochemical constants determined for the powdered aerial parts of L. dentata L. and L. angustifolia Mill. (Lamiaceae), cultivated in Egypt:

Physiochemical	Mean (%W/W)								
parameter	Description	L. dentata L.	L. angustifolia Mill.						
Moisture content	Loss on drying at 110 °C	15.28 ± 2.30	13.40±2.25						
Ash values	Total ash	10.48±2.33	8.55±1.96						
	Acid insoluble ash	1.90 ± 0.85	1.70±0.80						
	Water soluble ash	2.70±1.49	2.69±1.53						
	Water insoluble ash	5.80±1.36	4.10±1.16						
Extractives by	Petroleum ether	6.22±1.75	5.90±1.68						
cold maceration	Ethyl acetate	$5.90{\pm}1.48$	5.50±1.51						
	Ethanol	9.06±1.90	8.90±1.88						
	Aqueous	10.65±2.15	10.05 ± 2.04						

Results tabulated representing the means of three successive readings ± standard error.

The percent yields of the hydro-distilled volatile oils from aerial parts of *L. dentata* L. and *L. angustifolia* Mill. Were 3.47 and 2.95 respectively, physical examination revealed that they share in having pale yellow colour and aromatic characteristic odour but differs in their specific gravities and refractive indices, table (3).

GC/MS analysis of the prepared volatile oils exhibited qualitative and quantitative variations tables, 4, 5 where twenty-six and thirty-one compounds were identified representing 96.43 % and 98.56 % of the total individual composition of *L. dentata* L. and *L. angustifolia* Mill. volatile oils respectively (**54** and **55**).

The percentage of oxygenated compounds of both oils were much higher (83.56% and 96.83%) than those of

hydro-carbon compounds (10.09 % and 1.73 %) where the percentage of oxygenated monoterpene compounds were (50.04 % and 84.52 %) represented by ten and twenty one compounds, mainly menthe-1,5-dien-8-ol <Para> (26.80%), 1,8-cineole (4.98%), geraniol (3.92%) and linalyl acetate (18.99%), citronellol <*Alpha*> (17.36%), menthe-1,5-dien-8-ol <*Para*> (16.21%), meanwhile, the percentage of oxygenated sesquiterpene compounds were (33.52% and 12.31%) represented by three and two compounds, mainly caryophyllene oxide (16.40%), guiaol <*Alpha>* (15.36%) and farnsol (12.22%) in the volatile oils of L. dentata L. and L. angustifolia Mill. respectively (54 and 55).

 Table 3: Percent yields, physical characters, specific gravities and refractive indices of the volatile oils of L. dentata

 L. and L. angustifolia Mill. (Lamiaceae), cultivated in Egypt:

Item	L. dentata L.	L. angustifolia Mill.
Percent vield	3.47	2.95
Colour	Pale yellow	Pale yellow
Odour	Aromatic	Aromatic
Specific gravity	0.88942	0.89320
Refractive index	1 51840	1 52135

Item	L. dentata L.	L. angustifolia Mill.
Total hydrocarbons	10.09	1.73
- Monoterpenes	1.53	1.11
- Sesquiterpenes	8.56	0.62
Total oxygenated compounds	83.56	96.83
- Oxygenated monoterpenes	50.04	84.52
- Oxygenated sesquiterpenes	33.52	12.31
Others	2.78	0
Total identified compounds	96.43	98.56

 Table 4: Percent identified chemical constituents of volatile oils of L. dentata L. and L. angustifolia Mill.

 (Lamiaceae), cultivated in Egypt:

The percentage of monoterpene hydrocarbon compounds were (1.53 % and 1.11 %) represented by two and three compounds, mainly tricyclene (1.03%), pinene *<Alpha>* (0.50%) and cymene *<Para>* (0.82%), pinene *<Alpha>* (0.17%) while the percentage of sesquiterpene hydrocarbon compounds were (8.56 % and 0.62%) represented by seven and five compounds, mainly aromandrene *<Allo>* (2.07%), humulene *<Alpha>*(1.94%), germacrene-D (1.48%) and bisaboline *<Beta>* (0.40%), cadenine *<Gamma>* (0.15%) in the volatile oils of *L. dentata* L. and *L. angustifolia* Mill. respectively (**54** and **55**).

In conclusion, the volatile oil of *L. angustifolia* Mill. was richer in its content of total oxygenated compounds (96.83%) compared to *L. dentata* L. volatile oil (83.56%) whereas its main constituents were of monoterpene group while those of *L. dentata* L. were of sesquiterpene group, fortunately they are sharing in containing high percent of the monoterpene menthe-1,5-dien-8-ol<*Para>* at levels of 26.80% and 16.21% respectively, meanwhile, the total hydrocarbon compounds in the volatile oil of *L. dentata* L. was comparatively richer in its content (10.09%) compared to the volatile oil of *L. angustifolia* Mill. (1.73%) where its main constituents were of monoterpene group (8.56%).

The overall gained results exhibited significant qualitative and quantitative diversity in the chemical compositions of the volatile oils of *L. dentata* L. and *L. angustifolia* Mill. which are matched with the reported results in much previous research works (4 and 9-20).

Quantitative estimation of the total phenol content of the aqueous and methanol extracts of the tested species using Folin-Ciocalteu's reagent table 6, showed that the aqueous and methanol extracts of *L. angustifolia* Mill. contained higher percent of phenolic compounds (194.95±2.55 and 152.50±2.21) compared to those for *L. dentata* L. (188.50 ±2.07 and 167.10±2.30) mg GAE g⁻¹ respectively, the total phenols were measured by in terms of gallic acid equivalent (the standard curve equation is $y=0.05X\pm 0.0545$, $r^2= 0.9873$); while quantitative estimation of total

flavonoids of tested species using aluminum chloride reagent revealed that the aqueous and methanol extracts of *L. angustifolia* Mill. contains higher percent (98.83±2.68 and 119.95±1.87) compared to those of *L. dentata* L. (98.83± 2.68 and 119.95±1.87) mg QE g⁻¹ respectively, the total flavonoid contents of the extracts in terms of quercetin equivalent (the standard curve equation is $y = 0.0067X \pm 0.0132$, r² = 0.999).

Oxidative stress in diabetes mellitus revealed the reduction in the antioxidant status and glycation of proteins, inactivation of enzymes, and alteration in structural functions of collagen basement membrane (56), in addition, tissue damage resulting from oxidative stress has been implicated in the pathology of a number of disorder diseases such as cancer, inflammatory joint disease, cardiovascular diseases, cataract and could play a role in neurodegenerative diseases and ageing processes (57) whereas antioxidants are substances or nutrients which can prevent or slow the oxidative damage (58).

Acute toxicity studies revealed that no toxicity symptoms or death in the given doses so the LD_{50} of the tested oils and extracts are greater than 2000 mg kg⁻¹b.wt. which made them safe drugs, hence 50 and 100 mg kg⁻¹b.wt.were selected as therapeutic doses.

Oxidative stress is a condition of reduction in antioxidant enzymes like SOD and CAT; these enzymes play an important role in maintaining physiological levels of oxygen and hydrogen peroxide by hastening the dismutation of oxygen radicals and eliminating organic peroxides and hydroperoxides generated from inadvertent exposure to Streptozotocin which play an important role in reducing cellular stress (**59**). The activities of SOD and CAT were significantly decreased in liver and kidney tissues of diabetic control rats due to inadequacy of the antioxidant defenses in combating ROS mediated damage while treatment with different extracts and volatile oils increased the activity of these enzymes and help to control free radicals when compared to diabetic rats compared to the standard drug Glibenclamide, tables (7 and 8).

Page 3.

Peak		\mathbf{R}_{t}^{*}	Percent				
No.	Compound Name	(min.)	L. dentata L.	L. angustifolia Mill.			
Monot	erpene hydrocarbons						
1	Tricyclene	8.40	1.03				
2	Pinene < <i>Alpha</i> >	10.24	0.50	0.17			
3	Camphene	10.44		0.12			
4	Cymene <i><para></para></i>	10.47		0.82			
Oxvge	nated monoterpenes	10117		0102			
5	1,8-cineol	12.96	4.98				
6	Sabinene hydrate <i><cis></cis></i>	13.40		1.14			
7	Linalol oxide <i><cis></cis></i>	13.56		2.41			
8	Terpinolene	13.70		1.04			
9	Linalool	13.94		4.43			
10	2-menthenol < <i>Cis</i> >	14.20		0.11			
11	Camphor <i><dextro></dextro></i>	14.46		1.65			
12	Isoborneol	14.65		0.11			
13	Borneol	14.86		0.43			
14	Terpinen-4-ol < <i>Levo</i> >	14.94		0.89			
15	Terpinene< <i>Gamma</i> >	15.02	0.93	0.90			
16	Linalyl acetate	15.24		18.99			
17	Geraniol	15.65	3.92				
18	Nonanoic acid	16.88	3.72				
19	Isopulegol	17.40	0.75				
20	Bornyl acetate	17.52		0.37			
20	Isobornyl acetate	17.73		0.86			
22	Cuminyl alcohol	17.85	0.64	0.80			
23	Lavendulyl acetate	17.93		3.75			
23	Decanoic acid	17.93	2.35				
25	Neryl acetate	18.00	2.33	0.72			
<u>25</u> 26	Geranyl acetate	18.17		3.04			
20	Cedrene< <i>Alpha</i> >	18.54		7.40			
27	Citronellol< <i>Alpha</i> >	18.75	2.96	17.36			
<u></u> 29	Mentha1, 5 dien-8-ol <i><para< i=""> ></para<></i>						
	Dihydrocarvyl acetate	19.30	26.80	16.21			
30		20.67	2.99	1.83			
31	Carvone	21.93	2.99				
	terpene hydrocarbons	22.70	1 1 1	0.07			
$\frac{32}{22}$	Caryophyllene <i><beta< i=""> ></beta<></i>	22.79	1.11	0.07			
33	Humulene < <i>Alpha</i> >	23.58	1.94				
34	Aromandrene < <i>Allo</i> >	24.60	2.07				
35	Germacrene-D Bicyclogermacrene	25.95	1.48	0.06			
36	2 12	27.38	0.06	0.02			
37	Bisaboline <i><beta< i=""> ></beta<></i>	28.09	1.05	0.40			
38	Cadenine <i><gamma></gamma></i>	28.41	0.85	0.15			
	nated sesquiterpenes	20.19	16.40	0.00			
<u>39</u>	Caryophyllene oxide	29.18	16.40	0.09			
40	Guaiol < <i>Alpha</i> >	30.48	15.63				
<u>41</u>	Farnsol	31.95	1.76	12.22			
<u>Others</u>		20.21	0.40				
42	Octadecatrenoic acid methyl	30.21	0.49				
	ester						
43	6-dodecanone	33.72	0.68				
44	Docosane	44.22	0.58				
45	17-octadecynoic acid	46.86	1.03				

 Table 5: Chemical composition of the volatile oils isolated from the aerial parts of L. dentata L.and L. angustifolia

 Mill. (Lamiaceae), cultivated in Egypt:

* R_t: retention time in minutes

 Table 6: Total phenol and flavonoid contents of aqueous and methanol extracts prepared from L. dentata L. and L. angustifolia Mill. (Lamiaceae), cultivated in Egypt:

	L. den	tata L.	L. angustifolia Mill.			
Item	Aqueous	Methanol	Aqueous	Methanol		
Total phenol (mg GAE g ⁻¹)	188.50 ± 2.07	167.10±2.30	194.95±2.55	152.50±2.21		
Total flavonoid (mg QE g ⁻¹)	90.40±1.57	116.15±1.91	98.83±2.68	119.95±1.87		

The tabulated results representing means ± standard errors.

CONCLUSION:

Vitamin C plays a central role in the antioxidant protective system, protecting lipids oxidation and diminishing the number of apoptotic cells, regenerating the oxidized vitamin E and acts as a non-enzymatic antioxidant (**60**). Diabetes result in significant decrease in vitamin C levels compared to control rats where its level was significantly restored in liver and kidney tissues of treated groups compared to the standard drug Glibenclamide, tables (7 and 8).

GSH has a multifaceted role in antioxidant defense; it is a direct free radical scavenger; act as co-substrate for peroxide detoxification. Diabetic oxidative stress decrease GSH level in liver and kidney tissues compared to control where significant elevation of GSH levels were observed in treated groups compared to the standard drug Glibenclamide, tables (7 and 8).

In conclusion, the gained results proved quantitative variation in the estimated physiochemical parameters, qualitative and quantitative diversity of volatile oils' physical characters, constants and chemical composition, in addition to quantitative variation in total phenol and flavonoid contents of *L. dentata* L. and *L. angustifolia* Mill. (Lamiaceae) cultivated in Egypt meanwhile concerning antioxidant studies; both plants proved to be of promising therapeutic value in preventing diabetic oxidative stress which can be attributed to their significant quenching impact on the extent of lipid peroxidation along with, enhancement of antioxidant defense systems in liver and kidney tissues.

Table 7: Results of enzymatic and non-enzymatic antioxidant potentials of the volatile oils and methanol extracts of *L. dentata* and *L. angustifolia* at two dose levels (50 and 100 mg/kg b.wt.) and Glibenclamide (0.025 mg/kg b. wt.) on liver tissues of adult albino rats:

Antioxidant Parameters		Groups										
				CID	Volatile oils			-	Methanol extracts			
		Normal control	Negative control	GLB	L. dentata		L. angustifolia		L. dentata		L. angustifolia	
					50 mg	100 mg						
	e a	7.119	4.074	6.310	5.735	5.984	6.112	6.346	6.877	6.949	6.982	7.502
	SOD U/g tissue	±	±	±	±	±	±	±	±	±	±	±
S	E I S	0.089	0.152	0.152	0.068	0.088	0.050	0.098	0.100	0.145	0.040	0.074
Enzymatic	CAT μ mole H ₂ o ₂ utilized/min/ mg protein	58.466 ± 0.273	43.140 ± 0.121	53.871 ± 0.199	46.323 ± 0.134	47.365 ± 0.193	50.023 ± 0.084	51.295 ± 0.125	55.577 ± 0.358	56.699 ± 0.210	57.288 ± 0.181	58.701 ± 0.223
	C resh Ie	1.794	0.765	1.441	0.986	1.157	1.398	1.535	1.758	1.876	1.898	1.876
Non-enzymatic	Vit. C mg/g fresh tissue	± 0.013	$\overset{\pm}{0.020}$	± 0.016	± 0.016	± 0.061	± 0.041	± 0.035	0.022	± 0.025	± 0.035	± 0.027
n-en	н ю.н	47.115	29.543	43.416	24.333	30.242	36.664	39.544	43.965	46.076	45.946	47.993
Noi	GSH µg/mg protein	± 0.141	± 0.181	± 0.178	± 0.128	± 0.160	± 0.553	± 0.212	± 0.158	± 0.195	± 0.085	± 0.171

The tabulated results representing the means \pm standard error, n = 10

Table 8: Results of enzymatic and non-enzymatic antioxidant potentials of the volatile oils and methanol extracts of *L. dentata* and *L. angustifolia* at two dose levels (50 and 100 mg/kg b.wt.) and Glibenclamide (0.025 mg/kg b. wt.) kidney tissues of adult albino rats:

						G	roups					
Antioxidant Parameters		Namel Naméra CLD			Volatile oils				Methanol extracts			
		Normal control	Negative control	GLB	L. der	ıtata	L. angu	ıstifolia	L. de	ntata	L. angı	ıstifolia
					50 mg	100 mg	50 mg	100 mg	50 mg	100 mg	50 mg	100 mg
	0 e	2.321	0.890	1.942	1.138	1.215	1.319	1.464	1.868	1.983	2.129	2.416
	SOD U/g tissue	± 0.080	± 0.030	± 0.037	± 0.047	± 0.019	± 0.029	± 0.104	± 0.052	± 0.042	± 0.035	± 0.125
Enzymatic	CAT M mole H ₂ 0 ₂ utilized/min/ mg protein	$29.276 \\ \pm \\ 0.095$	15.679 ± 0.185	27.908 ± 0.268	18.710 ± 0.159	20.783 ± 0.171	21.142 ± 0.234	22.579 ± 0.162	24.938 ± 0.076	25.942 ± 0.076	28.518 ± 0.231	30.799 ± 0.201
Non-enzymatic	Vit. C mg/g fresh tissue	$1.270 \\ \pm \\ 0.024$	$0.495 \\ \pm \\ 0.024$	1.676 ± 0.025	$0.832 \\ \pm \\ 0.016$	0.889 ± 0.017	$0.887 \\ \pm \\ 0.020$	$1.064 \\ \pm \\ 0.038$	$1.210 \\ \pm \\ 0.010$	$1.211 \\ \pm \\ 0.038$	1.292 ± 0.016	1.431 ± 0.017
Non-en	GSH μg/mg protein	48.638 ± 0.297	23.453 ± 0.197	$44.786 \\ \pm \\ 0.155$	32.870 ± 0.116	34.895 ± 0.162	$32.848 \\ \pm \\ 0.335$	35.493 ± 0.219	40.835 ± 0.152	$\begin{array}{c} 43.589 \\ \pm \\ 0.205 \end{array}$	42.735 ± 0.185	46.683 ± 0.139

The tabulated results representing the means \pm standard error, n = 10

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Page 37

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Page 35