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RESEARCH ARTICLE

Survey on Diabetes and Standardization of Polyherbal Formulation

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ABSTRACT

Standardization of polyherbal formulation is important to validate the quality of drugs and to ensure that the consumers are getting medication which guarantees purity, safety, potency and efficacy. The present paper reports standardization of traditional ayurvedic liquid polyherbal antidiabetic formulation (Sucrogen) and diabetes survey for retrieving the information on medication along with the lifestyle of diabetic population. Sucrogen was standardized based on ayurvedic pharmacopeia physico-chemical properties, preliminary phytochemical tests, organoleptic characters, stability studies, microbial studies, TLC, HPLC, heavy metal estimation by AAS and flame photometry to fix the quality standard of this drug. Invitro anti-diabetics activity of the drug was determined using alpha amylase Inhibitory method. These studies resulted in a set of diagnostic characters essential for its standardization. The phytochemical constituents found to be present in raw materials used for the preparation of Sucrogen possibly helps in achieving the desirable therapeutic efficacy of the ayurvedic formulation.

KEYWORDS: Sucrogen, ayurvedic polyherbal formulation, standardization, survey

OBJECTIVE

To do a general survey on diabetes and to standardize the polyherbal formulation consumed by the population of Jharkhand for the determination of purity and quality of drug

INTRODUCTION:

levels in body increases due to either insufficient insulin as field survey from hospitals and healthcares. The online production or utilization. There are different types of link www.diabetessurvey.wordpress.com was made and diabetes namely type 1 diabetes, type 2 diabetes, was shared among our friends and relatives through mails gestational diabetes, diabetes insipidus, canine diabetes and social network sites. Based on the result obtained we feline diabetes etc. Here we are dealing with type 2 got to know the percentage of population having diabetes diabetes where the body produces plenty of insulin but the and whether they are undergoing ayurvedic, allopathy or cells are unable to utilize it. According to the year 2010 in homeopathy treatment and simultaneously ensured them India 50.8 million people have acquired the disease and the efficacy and purity of oral herbal drug. according to International Diabetes Federation (IDF) it is estimated that by year 2030, 80.4 million people will STANDARDIZATION: acquire this disease. As per World Health Organization (WHO) 50% population consume herbal drug for diabetes. standardization parameters studied were organoleptic, Oral antidiabetic drugs and insulin therapy are available for physicochemical properties, determination of pH, its treatment, but a trend of shifting to herbal drugs have fluorescence analysis, preliminary phytochemical tests, been seen among people due to its low cost, effectiveness density, viscosity, qualitative analysis by HPLC, TLC, and less side effects. So in this modern medicine era it is microbial studies and estimation of metals by flame important to standardize traditional drugs to justify its photometry, AAS. The organoleptic tests to determine the acceptability. The polyherbal formulation i.e. Sucrogen was color, odor, taste and texture of the drug. Fluorescence available from the market. It is manufactured by Joysree studies will be carried to determine the characteristics Kutir Silpam, Deoghar, Jharkhand and consumed by under UV lamp. Jharkhand population.

SURVEY:

A general survey on diabetes in Jharkhand was which include: done which gave us the idea of the medication and lifestyle

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Diabetes is a chronic disease where the glucose followed by people of Jharkhand. It included online as well

To assess the quality of the formulation various

MATERIALS AND METHODS:

The formulation Sucrogen contains 13 ingredients

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Sanskrit/Hindi name	Biological name	Family	Each 5ml contains	
Methi	Trigonella foecum graceum	Fabaceae	200 mg	
Jamun beej	Syzgium cumini	Myrtaceae	200 mg	
Satavari	Asparagus racemosus	Asparagaceae	100 mg	
Aswagandha	Withania somnifer	Solanaceae	100 mg	
Neem	Azadirachta indica	Meliaceae	100 mg	
Guduchi	Tinospora cordifolia	Menispermaceae	100 mg	
Gokshura	Tribulus terrestris	Zygophyllaceae	50 mg	
Harida	Curcuma longa	Zingiberaceae	50 mg	
Nayantara	Catharanthus roseus	Apocynaceae	50 mg	
Piashal	Terminalia tomentosa	Combretaceae	50 mg	
Triphala/Haritaki Vibbitaki	Terminalia chebula,	Combretaceae,		
	Terminalia bellirica,	Combretaceae,	50 mg	
Allididkij	Phyllanthus emblica	Phyllanthaceae		
Trikatu(Pippali,Maricha,	Piper longum, Piper	Piperaceae, Piperaceae,	50 mg	
Sunthi)	nigrum, Zingiber officinale	Zingiberaceae		
Kalamegha	Andrographis paniculata	Acanthaceae	50 mg	

Table No. 1The formulation Sucrogen contains 13 ingredients which include:

All tests were performed following standard protocols. **ORGANOLEPTIC TESTS:**

The color, odor, taste and texture were reported taking 5 ml of sample.

pH:

The pH test was performed by the pH paper and by pH meter method. Took the sample in a beaker and added a drop on the pH paper. The color change was compared with the standard and reported. For pH meter method, standard solution of pH 4 was taken. Then immersed the (107°C), cool and cold temperature. electrode in the beaker containing sample and noted the observed pH.

DENSITY:

The density of the formulation was found by the specific gravity bottle method. A clean dry specific gravity FLUORESCENCE ANALYSIS: bottle was taken and empty weight was taken using (w_1) Filled the bottle with distilled water and took the weight determined with different chemical reagents under (w_2) . Then filled the bottle with the sample and took the weight (w₃). The density was calculated using the formula

Density = $w_3 - w_1 \div w_2 - w_1$

VISCOSITY:

Viscometer was used to determine the viscosity. A stock solution of 1% was made and from that 0.1, 0.2 and METAL ANALYSIS: 0.3% concentration of a solution was made. The viscosities of these solutions were found concordant readings by the ESTIMATION OF CALCIUM, AND POTASSIUM BY FLAME formula

$\rho = t_{s \pm} t$

Where, t is the time taken by water and t_s is the time taken by sample

SOLUBILITY:

The solubility in various polar and non polar solvents was observed. The solvents included were water, ethanol, methanol, petroleum ether, hexane, benzene and chloroform.

STABILITY STUDIES:

The drug was subjected to stability testing by keeping at room temperature, accelerated temperature

PRELIMINARY PHYTOCHEMICAL TESTS:

The preliminary phytochemical qualitative tests were done following standard protocols.

Fluorescence characters of the sample were fluorescence, short range UV, long range UV and daylight. The sample was taken on a watch glass and treated with various reagents for the presence of fluorescence characters.

PHOTOMETRY:

Made a stock solution of 200 ppm of each metal to be analyzed and diluted to solutions of 100, 80, 60, 40 and 20 ppm. First calibrated with the standard solutions taking

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dilution factor as 1 and then estimated the sample. From Injection volume: 20µl the graph obtained reported the concentration of calcium, Detection at: 230nm and potassium.

ESTIMATION OF ZINC, MAGNESIUM AND IRON BY **ATOMIC ABSORPTION SPECTROMETER:** NITRIC ACID DIGESTION:

The sample was subjected to nitric acid digestion a day before. The sample and ml of concentrated nitric acid developed in iodine chamber and observed in UV chamber. in a boiling tube was subjected to a temperature of °20 for 45 minutes and then added ml of concentrated nitric MICROBIAL STUDIES: acid. Kept the boiling tube on sand bath at a temperature of 150°C for 8 hours till the sample becomes clear.

PREPARATION OF STANDARD SOLUTIONS: Zn (100 ppm):

 $ZnSO_4$ and $1ml 5M CH_3COOH$ in 100 ml of distilled water. control. Observed the plates for growth of microorganisms. Dilutions of 20, 40, 60 and 80 were made.

Mg (10 ppm):

0.5 g and concentrated HCl to dissolve and made up the volume to 100 ml. Dilutions of 2, 4, 6, 8 was made. Fe (10 ppm):

upto 100 ml. Dilutions of 2, 4, 6, 8 was made.

The calibration with standard solutions and estimation of added 10 ml of distilled water and measured the the metal concentration in sample was done using Varian absorbance at 540 nm. AA 240 AAS

HPLC (Waters 1525 binary HPLC pump):

water in the ratio 50:50 was used. Flow rate: 1ml/min **RESULTS:**

Column used: C18 TLC:

Made silica plates and added the sample 2cm from the edge of plate. Saturated the walls of the beaker with the solvent system. Kept the plates in the mobile phase of methanol and water (50:50). Then the plates were

The formulation was subjected to microbial test to ensure that it is free from microorganisms. The petri plates, swabs, nutrient agar media was autoclaved at 121C for 30 minutes. The media poured in the petri-plate and allowed to solidify. Inoculated the media with the formulation. Kept Made the stock solution by dissolving 0.44 g of for incubation at 3% for 24 hours keeping one plate as

AMYLASE INHIBITORY TEST

500µl of sample and 500µl of phosphate bufferalpha amylase incubated at 26 for 10 minutes. Adde d 500µl of 1% starch solution and incubate for 20 minutes at 0.7022 g of FAS and 2.5 ml of 1M H₂ SO₄ and made 25°C. 1 ml of DNS reagent was added and kept in boiling water bath for 5 minutes. Cooled at room temperature, % of amylase inhibitory = (Absorbance of solvent with enzyme- absorbance of solvent) - (absorbance of sample-absorbance of blank) ÷ Mobile phase: A solvent system of methanol- (Absorbance of solvent with enzyme- absorbance of solvent x 100



ORGANOLEPTIC PROPERTIES

Color: dark brown Odor: bitter

Taste: bitter

Texture: fine

рΗ

The pH was found to be **5 to 6** by pH paper and **5.8** by pH meter.

VISCOSITY:

DENSITY

Weight of empty bottle $(w_1) = 25.1305g$ Weight of water + bottle $(w_2) = 52.1811g$ Weight of sample + bottle $(w_3) = 52.8514g$ Density= w_3 - w_1 ÷ w_2 - w_1 = 52.8514-25.1305 ÷ 52.1811-25.1305 = **1.02 g/ml**

Concentration (%)	Trial 1 (sec)	Trial 2 (sec)	Trial 3 (sec)	Concordant reading (sec)
0.1	1.40	1.41	1.41	1.41
0.2	1.44	1.43	1.44	1.44
0.3	1.47	1.47	1.48	1.47
Water	1.59	1.58	1.59	1.59

Viscosity= Time taken by sample ÷ Time taken by water

Viscosity= Time taken by sample ÷ Time taken by water	Viscosity= Time taken by sample ÷ Time taken by water
Viscosity= Time taken by sample ÷ Time taken by water	Viscosity= Time taken by sample ÷ Time taken by water
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Water	Soluble
Ethanol	Soluble
Methanol	Soluble
Petroleum ether	Not Soluble
Hexane	Not Soluble
Benzene	Not Soluble
Chloroform	Not Soluble

STABILITY STUDIES:

The formulation was found stable at room, accelerated and cold temperature.

PRELIMINARY PHYTOCHEMICAL TESTS:

Alkaloids	+
Glycosides	-
Saponins	+
Carbohydrates	-
Sterols	-
triterpenoids	+
Flavanoids	+
Proteins and amino acids	+
Tannins	+

FLUORESCENCE ANALYSIS:

Sample+reagent	Day light	Fluorescence	Short UV	Long UV
Conc. HCl	Brown	Brown	Yellow	Green
Conc. HNO ₃	Brown	Yellowish brown	Purple	Yellowish green
CCI ₄	Brown	Brownish yellow	Greenish yellow	Yellowish green
CH ₃ COOH	Brown	Brown	Purple	Green
NH ₃	Brown	Brownish yellow	Yellow	Greenish yellow
FeCl ₃	Brownish yellow	Reddish yellow	Bluish yellow	Yellow

METAL ANALYSIS:

Metals	Concentration
Potassium	15 ppm
Calcium	4 ppm
Zinc	1.4648 ppm
Magnesium	0.4415 ppm
Iron	0.0207 ppm



Peak Results						
	Name	RT	Area	Height	Amount	Units
1		1.522	12293094	982133		
2		1.921	9638480	404275		
3		3.769	7685627	286263		
4		5.294	3624705	109824		

TLC:

The spots were obtained at Rf value 0.47 and 0.63 which corresponds to the standard value of 0.5 for rutin ⁽⁵⁾ and 0.66 for diosgenin⁽⁶⁾.

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MICROBIAL STUDIES: No growth was seen for the plate inoculated with drug.



Petriplate inoculated with drug showed no growth of microorganisms

AMYLASE INHIBITORY TEST:

The sample showed inhibition of amylase enzyme with increase in concentration.

Concentration	% of amylase inhibition	
Standard	0.1518	
Test solution 3(3%)	0.1480	
Test solution 2(2%)	0.1591	
Test solution 1(1%)	0.1682	
Solvent with enzyme	0.0761	
Solvent (water)	0.0390	
Blank	0.1447	
Sample	Absorbance	

Concentration	% of amylase inhibition
1%	36.65
2%	61.18
3%	91.10
Standard (Mannitol)	80.8



DISCUSSION:

that majority of population are consuming ayurvedic drug significant phenomenon exhibited by various chemical Sucrogen which is effective and does not possess any side constituents in plants. The substances can be derivatized effects. The polyherbal formulation Sucrogen was by reagents if they are not themselves fluorescent hence subjected to various analytical techniques. organoleptic parameters revealed that dark brown color, way and it is an important parameter of pharmacognostical bitter taste and odor and fine texture of the formulation. evaluation. Metal estimation showed the presence of The pH was found to be between 5 and 6. The solubility in potassium, calcium, zinc, magnesium and trace amount of water, ethanol and methanol shows that it is soluble in iron. HPLC result showed a peak at 1.921, 3.769 and 5.294 polar solvents. The phytochemical tests are in accordance minutes which corresponds to the peak of nimbidin, rutin, with the ayurvedic pharmacopeia standards. Presence of and diosgenin respectively which is in accordance with the

tannins and alkaloids were the supporting evidence for From the survey done in Jharkhand, it was found antidiabetic property of Sucrogen. Fluorescence is a The some crude drugs are often assessed qualitatively in this

standard values^(7,8,9). Nimbidin is an active constituent of antioxidative effect of dietary diosgenin and chromium neem. Diosgenin, a saponin is found in Methi. Rutin is a chloride supplementation on high -cholesterol fed flavanoid found in jamun beej. All these three components Japanese quails, Pakistan journal of biological sciences, possess antidiabetic property. In TLC Rf values are obtained 14(7) 425-432. at 0.47 and 0.63 respectively for rutin and diosgenin. The 7. plate inoculated with the drug showed no microbial growth Somnath bhadury, Tanusree nag and Simli sarkar (2009), hence the drug is free from contamination. The amylase Bioremediation of heavy metals from neem(Azadirachta enzvme inhibition was observed with concentration of drug hence antidiabetic property of the journal of pharmaceutical and clinical research, volume 2 formulation was confirmed invitro.

CONCLUSION:

standardized as per the WHO guidelines and ayurvedic international Journal of Pharma and biosciences. Vol 1(2). pharmacopeia. Based on the survey as well as 9. standardization done it has been found that Sucrogen has Ismail and Nibal D. El-Tanbouly, (2006), Biochemical study good antidiabetic property and can be used for evaluating of antidiabetic action of the Egyptian plants fenugreek and the quality and purity of formulations for polyherbal drug.

REFERENCE:

1. Bisht, (2011), Phytochemical screening and standardization journal of traditional knowledge, vol7(3), pp.389-396. of polyherbal formulation for dyslipidemia, International 11. journal on pharmacy and Pharmaceutical Sciences, Vol 3 N.Kamalakkannan, (2006), Supp 3.

2. Pradeep Kumar Samal, (2008) standardization strategies of *biochemical and molecular toxicology*, vol 20,96-102. herbal drugs-an overview Research J. Pharma and Tech 12. 1(4):page 310-312.

3. Kartik Ch patra, K Jayaram Kumar, P Suresh, (2009) saranya,(2011), standardization of polyherbal siddha formulation amukkara phytochemical study of Lagerstroemia speciosa leaves, choornam, International journal of traditional knowledge international journal of research in ayurveda and Vol 8(3), pp.449-452.

Soni Hardik K, Ribadiya Nikunj, Bhatt Surendra B, 13. 4. Sheth Navin R, (2010) Evaluation of herbal formulation formulation parameters of an antistress herbal capsule, (capsule) containing ashwagandha as a single herb with *international journal of Pharma and biosciences*,vol2. their nutritional value determination, International Journal 14. of Applied Biology and Pharmaceutical Technology, vol 1: Upendra issue 3.

5. chromatography methods for rapid identity testing of rhizome, International journal of current pharmaceutical Morinda citrifolia L.(Noni) Friut and leaf, *advance journal of research*, vol2 36-39. food science and technology 2(5): 298-302.

6. A.Oriquat, Mahmoud Abu-Samak, Khtan A.Al-Mzain and herbs and herbal drugs used for the treatment of diabetes. Maher Salim, (2011), The hypocholesterolemic and

Amitava ghosh, Piyali chakrabarti, Partha roy, increase indica) leaf extract by chelation with dithizone. Asian issue 1.

8. Willy shah, Nilan rane, M. B. kekare and Vikas vaidya, (2010), Estimation of two bioactive compounds Ayurvedic formulation Sucrogen has been from azadiracta indica A. juss. Leaves using HPLC,

> Mohamed Z. Gad, Maha M. El-Sawalhi, Manal F. balanites Molecular and Cellular Biochemistry 281: 173-183.

10. K R Gopala Simha, V.Laxminarayana, SVLN Prasad, Shahjahan Khanum, (2008) standardization of yogaraja N.V Sathesh Madhav, Kumud Upadhayaya, Asha gugullu- an ayurvedic polyherbal formulation, International

P.Stanley Mainzen prince, Rutin improves glucose homeostasis in streptozotocin diabetic tissues by altering Neeli Rose Ekka, Kamta Prasad Namdeo and glycolytic and gluconeogenic enzymes, Journal of

> Vijayaraghavalu Sai Saraswathi, Dhakshanamurthy Thirumalai, Pothula kushal Yadav and Muniswamy pharmacognostic and preliminary pharmacy, vol2(3) 893-898.

> Nasreen.S(2011), Evaluation of preformulation and

Arun kumar beknal, Prakash g korwar, M.A.Halkal, kulkari, Basawaraj S.Patil, Srinivas R.Soodam(2010), Phytochemical investigation and Brett J. West and Shixin Deng, (2010), Thin layer antioxidant activity study of drynaria quercifolia linn

15. Manisha modak, Priyanjali dixit, Jayant londhe, Saroj Hisham Y.Al-Matubsi, Nagham A.Nasrat, Ghaleb ghakadbi, and Thomas paul A.Devasagayam, (2006), Indian